

Old burned bones tell us about past cultures

G. Festa,^a C. Andreani,^{a-c} M. Baldoni,^{d,e} V. Cipollari,^f C. Martínez-Labarga,^{c,d} F. Martini,^g O. Rickards,^{c,d} M.F. Rolfo,^h L. Sarti,ⁱ N. Volante,ⁱ R. Senesi,^{a-c} F.R. Stasolla,^j S.F. Parker,^{k,*} A.R. Vassalo,^l A.P. Mamede,^l L.A.E. Batista de Carvalho^l and M.P.M. Marques^{l,m}

^aCENTRO FERMI – Museo Storico della Fisica e Centro Studi e Ricerche “Enrico Fermi”, Piazza del Viminale 1, 00184, Rome, Italy

^bUniversità degli Studi di Roma “Tor Vergata” – Dipartimento di Fisica, Via della Ricerca Scientifica 1, 00133, Rome, Italy

^cUniversità degli Studi di Roma “Tor Vergata” – Centro NAST, Via della Ricerca Scientifica 1, 00133, Rome, Italy

^dUniversità degli Studi di Roma “Tor Vergata” – Dipartimento di Biologia, Via della Ricerca Scientifica 1, 00133, Rome, Italy

^eUniversità degli Studi di Roma “Tor Vergata” – Dipartimento di Biomedicina e Prevenzione, Via Montpellier 1, 00133, Rome, Italy

^fSoprintendenza Archeologia del Lazio e dell’Etruria Meridionale, Via Pompeo Magno 2, 00192, Italy

^gUniversità degli Studi di Firenze – Dipartimento di Storia, Archeologia, Geografia, Arte e Spettacolo, Via S. Gallo 10, 50129, Firenze, Italy

^hUniversità degli Studi di Roma “Tor Vergata” – Dipartimento di Storia, Patrimonio culturale, Formazione e Società, Via Columbia 1, 00133, Rome, Italy

ⁱUniversità di Siena – Dipartimento di Scienze Storiche dei Beni Culturali, Via Val di Montone 4, 53100, Siena, Italy

^jSapienza Università di Roma, – Dipartimento di Scienze dell’Antichità, Piazzale A. Moro 5, 00185, Rome, Italy

^kISIS Facility, STFC Rutherford Appleton Laboratory, Chilton, Didcot, OX11 0QX, UK. E-mail: stewart.parker@stfc.ac.uk,  0000-0002-3228-2570

^lUniversity of Coimbra – Unidade de I&D “Química-Física Molecular”, Department of Chemistry, Coimbra 3004-535, Portugal.  0000-0002-8391-0055

^mUniversity of Coimbra – Department of Life Sciences, Coimbra 3000-456, Portugal.  0000-0002-8059-8537

Introduction

Bone is a composite biomaterial mainly comprised of packed collagen fibres and an inorganic matrix of crystalline hydroxyapatite [$\text{Ca}_{10}(\text{PO}_4)_6\text{OH}_x$, Hap], the hydroxyl and phosphate groups being partly substituted by carbonate. Bone mineral characteristics, such as crystallinity and carbonate/phosphate content vary as a function of age, sex, location in the skeleton, diet and pathological state of the person. With increasing burning temperature ($>200^\circ\text{C}$), the bone matrix suffers structural and dimensional variations, usually leading to higher crystallinity. These are reflected in the vibrational spectra and a combination of inelastic neutron scattering (INS), infrared and micro-Raman spectroscopies was used by the authors for the investigation of

human bones burned under controlled conditions.¹ This has enabled definitive conclusions about the temperature at which the bone was burned to be made. INS is particularly well-suited for studying hydrogenous compounds.² The neutron scattering cross-section (σ) of an atom is characteristic of that atom and independent of the chemical environment. Since the value for hydrogen ($\sigma=80$ barns) far exceeds that of all other atoms (typically $\sigma\sim 5$ barns), the modes with significant hydrogen displacement dominate the spectra.

Burned bones are often found in archaeological sites as a result of fire or funerary practices and are often the only preserved human remains. The specimens currently under study (both human and faunal) were found at different archaeological sites in Italy, from the Neolithic, the copper age, Roman period and the Middle Ages (Figure 1).

Interpretation of the diagenetic alterations in the analysed samples was based on vibrational spectroscopic results previously¹ gathered for modern human bones burned at well-defined temperatures (400–1000 °C). Hydroxyapatite’s hydroxyl libration (OH_{lib}) at $\sim 630\text{ cm}^{-1}$ in particular, was found to be a very useful diagnostic signal of the burning process, as it was found to undergo a clear shift to lower energy with increasing burning temperature. Comparison between these data and that measured for the archaeological samples has yielded valuable information on ancient practices and customs.

Roman bones

Figure 2 shows the INS spectra of human bones, from the same skeleton, discovered in tomb 36 at the *Guidonia-Montecelio* Roman archaeological site (Figure 1C). The skeleton was found

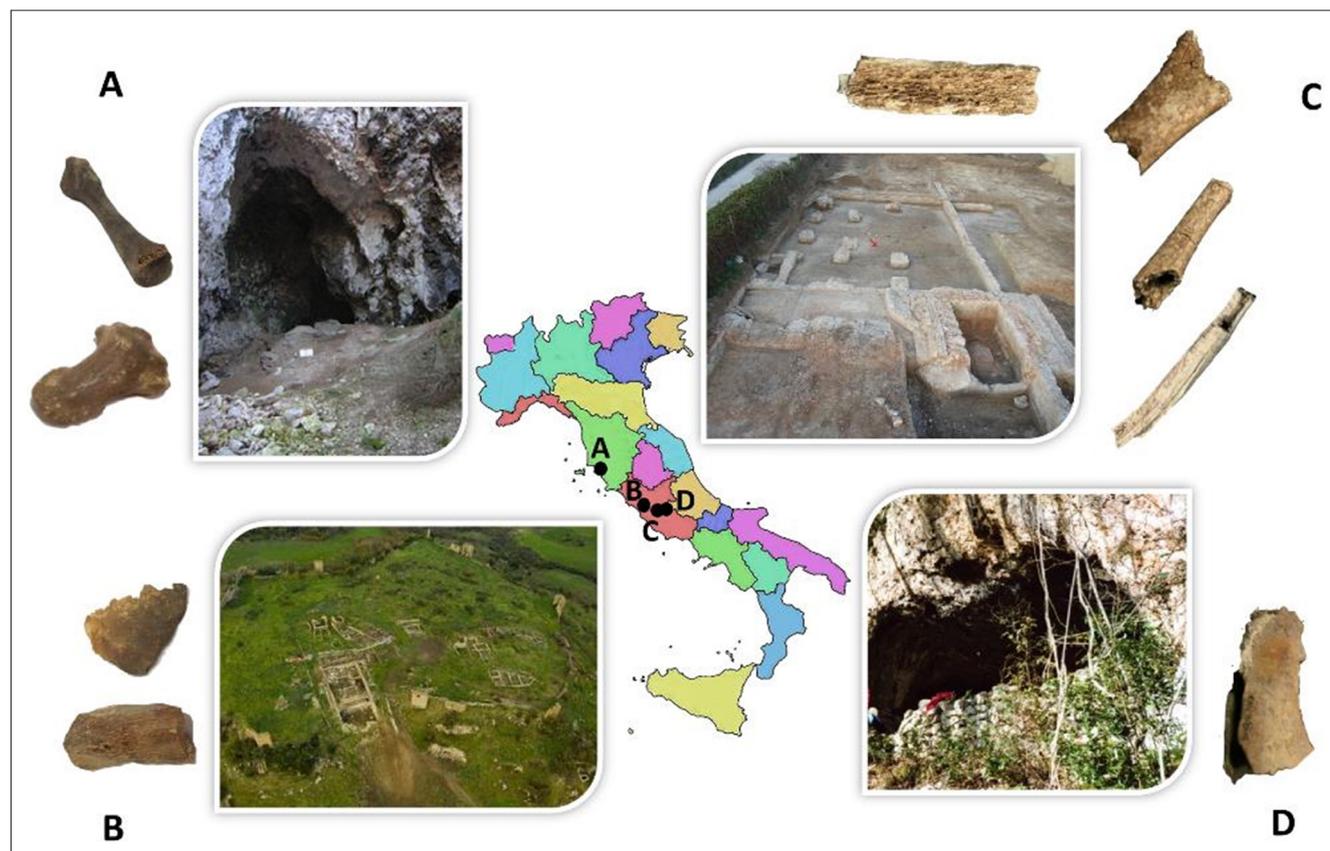


Figure 1. Archaeological sites where the samples investigated were found. A, *Scoglietto* cave—hand distal phalanx and metacarpal bone; B, *Leopold-Cencelle*—skull and tibia fragments; C, *Guidonia-Montecelio* (tomb 36)—fibula, ulna, femur and humerus fragments; D, *Mora Cavorso* cave—sheep/goat jaw fragment.

intact, incorporated in a clay matrix and placed supine with the arms along the pelvis. Surprisingly, different types of bone from the same skeleton displayed very different INS profiles, consistent with diverse heating conditions from below 400 °C up to 500 °C or 800–900 °C. This can be explained by ancient funerary practices that involved burning the corpse inside the grave. Correlating the heating conditions with the type of bone, it can be seen that the temperature progressively increases for the ulna, femur, humerus and fibula which indicates that the body may have been folded in a foetal position. In addition, for some of these samples a band at $ca\ 900\text{ cm}^{-1}$, ascribed to $\nu(\text{HCO}_3^-)$, was detected (mainly for the fibula), showing contamination from the soil which has a limestone composition in the *Guidonia-Montecelio* region (since the samples were found unwrapped inside the earth tomb). The major factors prone

to influence the degree of combustion of the skeletal remains, and, therefore, the effect on bone composition and crystallinity, are the temperature, the heating time and the environmental conditions (e.g. presence or absence of oxygen). The spectroscopic data depicted in Figure 2 show that the lower area of the skeleton reached higher temperatures than the upper part. Coupled with the discovery of lamp fragments by the feet of the body, it suggests that the burning occurred *in situ* and that the lamp had an active role in this ritual funerary practice.

Copper age bones

The infrared and INS spectra of human hand distal phalanx (PFF12) and metacarpal bone (PFF27) from the copper age, found at the *Scoglietto* cave, a 40 m deep cave located in the *Parco Naturale della Maremma* near Grosseto (Tuscany, Italy) (Figure 1A) are shown

in Figure 3. The FTIR-ATR results show that these samples were subject to heating at temperatures no higher than 450–500 °C. They display the typical infrared features of bone still containing traces of protein (collagen I), mainly evidenced by the amide I band (Figure 3A). The INS data goes further in identifying some lipid components through the corresponding $\delta(\text{CH}_2)$ and $\nu(\text{CH})$ signals at $ca\ 1450\text{ cm}^{-1}$ and 3000 cm^{-1} , respectively (Figure 3B). Upon comparison with the data obtained for reference samples heated at defined temperatures, both spectroscopic signatures correspond to a burning temperature around 500 °C, virtually homogeneous for the three ancient skeletal remains. This temperature is compatible with an incomplete cremation in home fires.

Raman micro-spectroscopy measurements performed on these ancient samples provided clear evidence of the presence of gypsum ($\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$),

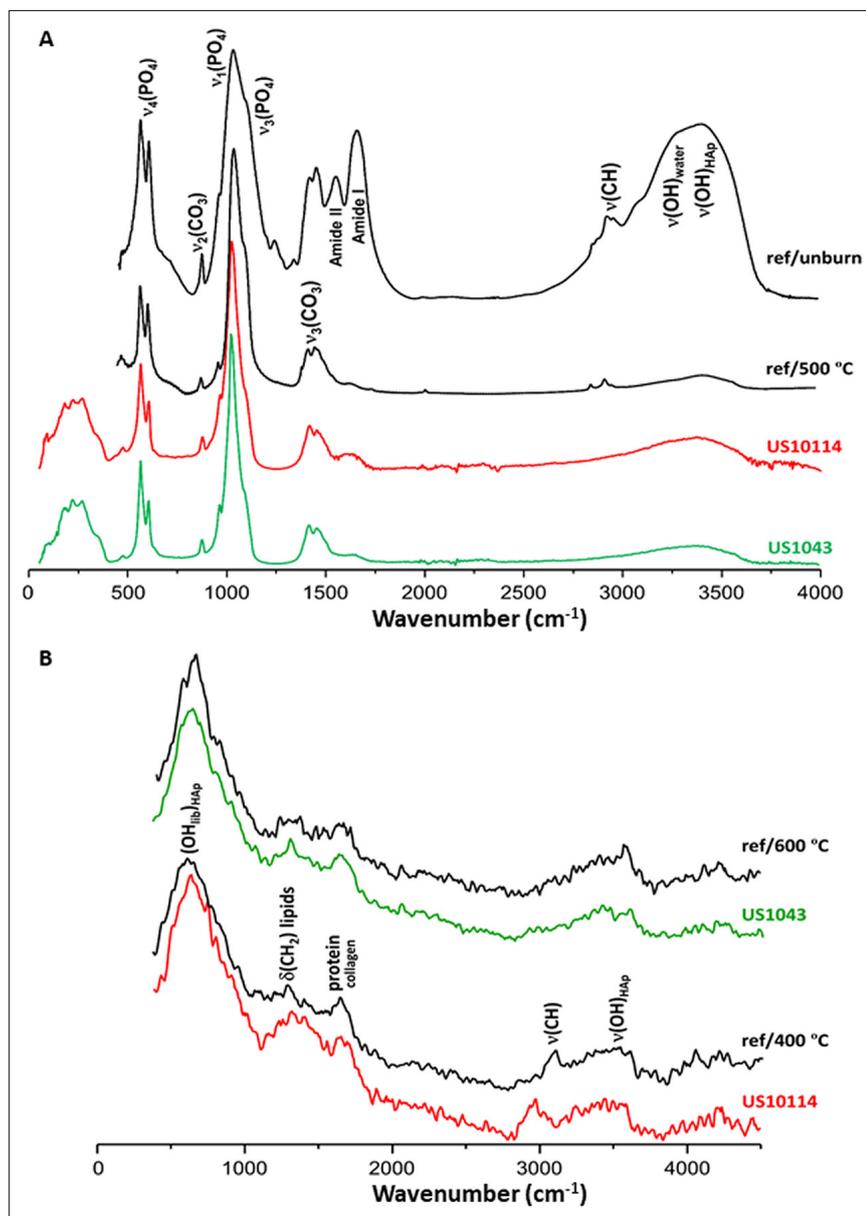


Figure 4. Vibrational spectra of medieval human bones (*Leopoli-Cencelle*, Italy). Comparison with human modern bone samples burned at 400, 500 and 600°C. A, FTIR-ATR; B, INS.

almost total loss of the organic constituents: the lipids having been completely destroyed above 400°C, while some collagen is still present within the bone matrix. Additionally, the detection of francolite revealed the occurrence of fluoride in the bone matrix, probably due to environmental contamination from either the neighbouring soil or water.

Neolithic bones

In order to assess the usefulness of this methodology and to extend its

capabilities beyond human skeletal remains, Neolithic faunal bones from the *Mora Cavorso* archaeological site (Figure 1D) were also investigated. By analysis of the most informative vibrational bands (from phosphate, carbonate and bone's organic components), the archaeological remains appear to have been exposed to temperatures below 500°C, which is compatible with the temperatures reached in hearths used in Neolithic settlements for cooking. However, care should be taken

when interpreting these data, since it was found that human and faunal bones display different reactions to heat, especially at higher temperatures.

Conclusions

This study (see Notice below) constitutes the first application of neutron spectroscopy to archaeological skeletal remains and will allow archaeologists/anthropologists to gather relevant information on ancient civilisations regarding the locations, and funerary, burial or cooking practices. Future research will enlarge the number of samples and archaeological sites in order to gather data on a wider range of contexts—geographical, historical and anthropological. Further spectroscopic biomarkers will then be identified, enabling a precise statistical analysis of the results linking particular spectral signatures to specific burning scenarios and environmental settings.

Notice

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